

NATIONAL UNIVERSITY OF SCIENCE AND TECHNOLOGY DEPARTMENT OF APPLIED CHEMISTRY BACHELOR OF SCIENCE HONOURS DEGREE SUPPLEMENTARY EXAMINATIONS – AUGUST 2010 INDUSTRIAL INORGANIC CHEMISTRY I SCH2114 TIME: 3 HOURS

INSTRUCTIONS TO CANDIDATES

Answer <u>ANY FIVE from the six provided.</u> Each question carries 20 Marks.

Total Marks - 100

1.	(a) List 4 physical properties of oxygen.	[4 marks]
	(b) Describe the cryogenic separation of oxygen from air.	[10 marks]
	(c) Name (6) industrial uses of Helium gas.	[6 marks]
2.	With the aid of a process flow diagram describe the contact manufacturing procacid.	ess of sulphuric [20 marks]
3.	(a) Account for the factors that affect the catalytic conversion of NH_3 from N_2 and H_2 . [6 marks]	
	(b) Explain the production of urea from dehydration of ammonium carbamate. A explain the likely hazards that go with this process.	Also [8 marks]
	(c) List 6 industrial applications of urea.	[6 marks]
4.	(a) List the advantages of the electric furnace method of phosphoric acid produc process method.	ction over the wet [4 marks]
	(b) Illustrate (3) three possible reactions that occur in the furnace during manufa phosphoric acid.	acture of [9 marks]
	(c) Describe (2) two methods by which the phosphoric acid can be purified	[4 marks]
	(d) Name (3) three uses of phosphoric acid.	[3 marks]
5.	(a) Describe the thermal properties of ceramics. For each property indicate an ir be utilised for its determination.	nstrument that can [12 marks]
	(b) Explain the process of devitrification in glass-ceramic products.	(4 marks)
	(c) Name (2) two applications of cermets.	[4 marks]

6.	(a) What is the chemical composition of Portland Cement?	[8 marks]
	(b) Describe the cement manufacturing process.	[12 marks]

End of Question Paper!