

**NATIONAL UNIVERSITY OF SCIENCE AND TECHNOLOGY**



**FACULTY OF SCIENCE AND TECHNOLOGY EDUCATION  
DEPARTMENT OF ART, DESIGN AND TECHNOLOGY EDUCATION  
MATERIAL SCIENCE**

**PDT2328**

**Main Examination**

**November 2024**

This examination paper consists of 4 pages

**Time Allowed: 3 Hours**

**Total Marks: 100**

**Examiner's Name: Mr T. Muzari**

**External Examiner: Dr.C.Kahanji**

**INSTRUCTION AND INFORMATION TO THE CANDIDATE**

1. Answer **Question 1(compulsory)** and any other three
2. Each question is worth 25marks
3. Use of calculators is permissible

**MARK ALLOCATION**

<b>QUESTION</b>	<b>MARKS</b>
<b>1</b>	<b>25</b>
<b>2</b>	<b>25</b>
<b>3</b>	<b>25</b>
<b>4</b>	<b>25</b>
<b>5</b>	<b>25</b>
<b>Total marks attainable by the candidate</b>	<b>100</b>

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### Question 1(Compulsory)

- a) Clarify the following terms considering that electrical conductivity depends on the number of electrons in the outer most shell:
- i. Conductors [2marks]
  - ii. Semi-conductors [2marks]
  - iii. Insulators [2marks]
- b) Briefly cite and describe the main differences between ionic, covalent, and metallic bonding. [9marks]
- c) Give a brief account of the classification of wood-based composite materials taking note of the raw materials and process types. [10 marks]

### Question 2

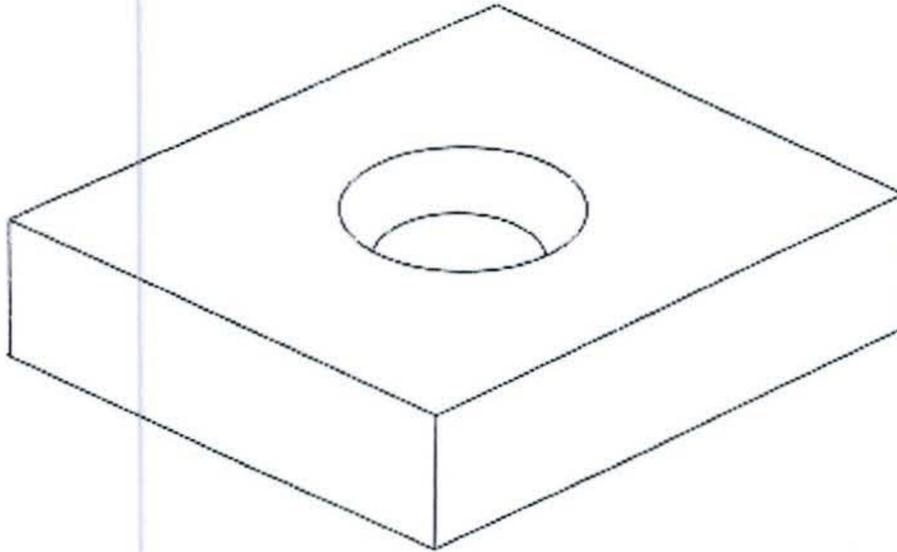
- a) Explain the difference between atomic structure and crystal structure. [6 marks]
- b) Sketch and show inter-axial angles and lengths of the following unit cells:
- i. Face-centred orthorhombic. [3 marks]
  - ii. body centred tetragonal. [3 marks]
  - iii. face centred cubic. [3 marks]
- c) Given an all right angled unit cell for a hypothetical metal measuring 0,35nm long, 0,35nm wide and 0,45nm high with an atomic weight of 141g/mol:
- i. State the crystal system this unit cell belongs to [2marks]
  - ii. What would this crystal structure be called? [2marks]
  - iii. Illustrate the unit cell showing the dimensions and angles respectively. [3marks]
  - iv. Calculate the density of the material giving your answer in grammes per cubic centimetres. [5marks]

### Question 3

- a) Describe the three main groups of engineering materials. [6marks]

b) **Figure Q3b** is a Block with a 200mm diameter through hole. Provide an explanation of making the hole if the block is made of :

- a) Wood [3marks]
- b) Metal [3marks]
- c) Plastic [3marks]



**Figure Q3b Block**

- c) Sketch three methods of dealing with sharp corners of the Figure 3Qb Block for increased user safety. [6marks]
- d) Assuming that Figure 3Qb Block is made of metal, describe two methods that **deteriorate** the block. [4marks]

**Question 4**

- a) Differentiate thermosetting and thermoplastic plastics. [4marks]
- b) Explain the reasons why some plastic materials deform with very low heat. [5marks]
- c) Describe **three** additives to polymers that impart special characteristics during polymer. [6marks]
- d) State and briefly explain any **five** methods used in forming and shaping thermoplastics. [10marks]

**Question 5**

- a) Describe the tripartite relationship of structure-property-processing for enhanced performance of engineering materials. [9marks]

- b) State and explain the factors engineers consider in the selection of engineering materials for an engineering application. **[6marks]**
  
- c) Distinguish the following principal properties of materials which are of importance to the designer:
  - i. Elasticity and plasticity. **[5marks]**
  
  - ii. Ductility and malleability. **[5marks]**

**END OF EXAMINATION**