# NATIONAL UNIVERSITY OF SCIENCE AND TECHNOLOGY

### FACULTY OF SCIENCE AND TECHNOLOGY EDUCATION

### ENVIRONMENTAL ENGINEERING

#### PTE2259

### Second Semester Examination Paper

May 2019

### This examination paper consists of 3 pages

Time allowed: 3 hours

Total Marks: 100

Special requirements: none

Examiner's Name: Mrs F. Makwiranzou

### **INSTRUCTIONS**

- 1) The paper consists of 3 printed pages with 5questions.
- 2) Each question carries 25 marks.
- 3) Answer any **4(four)** questions
- 4) Answer a new question on a fresh page.

## MARK ALLOCATION

QUESTION	MARKS
1	25
2.	25
3.	25
4.	25
5.	25
TOTAL	100

Page 1 of 3

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### **QUESTION 1**

- a) Define water pollution [2]
- b) List any five wastewater characteristics. [5]
- c) With the aid of a clearly labelled diagram describe the process of eutrophication. [8]
- d) List and describe fully the causes of water pollution. In your answer include ground water pollution and surface water pollution. [10]

### **QUESTION 2**

- a) Briefly discuss five objectives of wastewater treatment.[10]
- b) Compute the DWF and the average BOD concentration given the following design data
  No. of residential units 7550
  Water Consumption 350l/ca.day

Water Consumption	350l/ca.day	
Industrial effluent	3000m³/day	
BOD load	60g/ca.day	
BOD for industrial effluent	2500mg/l	[15]

### **QUESTION 3**

- a) Describe the biochemical processes that take place in nutrient removal in a treatment plant. [5]
- b) Draw a flow diagram indicating all stages and processes involved in wastewater treatment from preliminary to tertiary treatment. [10]
- c) Explain the objectives of preliminary treatment and describe the processes used to achieve these objectives. [10]

#### **QUESTION 4**

- a) From first principles derive the Phelps equation. [5]
- b) Differentiate between biological Oxygen Demand (BOD) and Chemical Oxygen Demand (COD). [10]
- c) Given the following parameters:

$$\begin{split} BOD_u &= 50mg/l\\ \Theta &= 1.05\\ K_{20} &= 0.23d^{-1} \end{split}$$

Temperature =  $35^{\circ}C$ 

Calculate **BOD**<sub>5</sub> for the wastewater [10]

### **QUESTION 5**

- a) Explain the common procedure used for the determination of the BOD for a given wastewater sample. [10]
- b) A BOD test is run using 100ml of treated wastewater mixed with 200ml of pure water. The initial DO of the mix is 9.0mg/l. After a long period of time, the DO is 2.0mg/l and it no longer seems to be dropping.
  - i. What is the 5 day BOD of the wastewater? [5]
  - ii. Estimate the ultimate BOD, neglecting the effects of nitrification. [5]
  - iii. What would be the remaining BOD after 5 days? [5]